## Environmental Concerns About Road Sait

## AMinnesota Polution Eofitrol-agency

Did you know that several metro Road salt is the most commonly lakes and streams in Minnesota have been classified as polluted by foad salt?

Winters in Minnesota bring slippery roads and the application of de-icing materials to keep our roads free from ice and safe. used de-icer, which contains sodium and chloride. The chloride in road salt enters our surface waters and groundwater after snow melts and is harmful to the fish, insects, and plants. The sodium stays in the soil and reduces its ability to retain water and increases the amount of erosion.

Too much salt results in costly damages and serious environmental consequences.

## While progress in Minnesota

 has been made in these areas, there is still much work to do in order to meet water quality standards and achieve a high level of road safety. In 2010, the MPCA more than doubled the number of waters that are listed as impaired for chloride.A study conducted by the University of Minnesota determined a chloride mass balance for the Twin Cities Metropolitan Area (TCMA) and found that approximately $78 \%$ of all chloride generated in the TCMA is being retained in the TCMA. This includes all of the main sources of chloride: chloride from road salt, wastewater treatment plants, water softeners, and other industrial sources.

Chloride is a conservative ion (meaning it moves with water without being broken down or lost). Once the chloride is in the water, the only known technology for its removal is reverse osmosis through massive filtration plants, which is not economically feasible. This means that chloride will continue to accumulate in the environ-
ment. A high chloride concentration in both the aquatic and terrestrial environment has some of the following implications for human consumption, aquatic life, and plant life:

- At high concentrations chloride is toxic to fish and insects
- At lower levels chloride can negatively affect the fish and insect community structure, diversity and productivity
- Direct road salt splash can kill plants
- Sodium in the road salt destroys soil stability, decreases the soils ability to infiltrate water, and can increase soil erosion.
- Some animals living near or relying on chloride polluted waters are sensitive to road salt While research does exist that identifies the negative impacts that chloride and specifically road salt has on the environment, there are still many unknowns. Continued research will help us to better understand how chloride interacts with the environment and therefore how to properly manage our water resources.

Follow these simple tips to protect our water!
There are many ways to reduce salt use while maintaining high safety standards.


Novotny et al. 2007, UMN

- Shovel. The more snow and ice you remove manually, the less salt you will have to use and the more effective it can be. Whether you use a shovel, snow blower, snow plow, or ice scraper, get out there as early as you can and keep up with the storm. You may even decide that salt isn't needed. - $15^{\circ} \mathrm{F}$ is too cold for salt. Most salts stop working at this temperature. Use sand instead for traction, but remember that sand does not melt ice. Use the reference table below to apply the correct product for the conditions.
- Slow down. Drive for the conditions and make sure to give plow drivers plenty of space to do their work.
- Be patient. Just because you don't see salt on the road doesn't mean it hasn't been applied. These products take time to work.
- More salt does not mean more melting. Use less than 4 pounds of salt per 1,000 square feet (an average parking space is about 150 square feet). One pound of salt is approxi-
mately a heaping 12 -ounce coffee mug. Consider purchasing a hand-held spreader to help you apply a consistent amount.
- Sweep up extra. If salt or sand is visible on dry pavement it is no longer doing any work and will be washed away. Use this salt or sand somewhere else or throw it away.
- Watch a video. This video, produced by the Mississippi River Watershed Management Organization, provides tips to homeowners about more environmentally friendly snow and ice removal: Improved Winter Maintenance: Good Choices for Clean Water. http://www.youtube.com/ watch? v=qc8Y-_Nmfmo
- Share a brochure. Read and pass along Nine Mile Creek Watershed District's brochure about residential snow and ice care. You can find it on Nine MIle Watershed District's education page. http://www.ninemilecreek. org/EDUCATION/EducationPrograms.asp
- Check out other resources. If you are responsible for snow and ice removal somewhere other than your home, please check out our training and resources tab.
Know about the salt product Salts can range from simple table salt to calcium chloride. Salts are used because they are able to decrease the freezing point of water. Whatever
product you chose, make sure you know at what temperature it stops working. We recommend using the table below as labels may be misleading. Note that pavement temperatures are usually warmer than air temperatures. To find out the pavement temperature near you, search the Road Weather Information Service, http:// www.rwis.dot.state.mn.us/.


Calibrate your equipment and know where you are applying your snow and ice removal chemistries. photo Ken Rost

## Winter Parking Lot and Sidewalk Maintenance

## Key Information Needed:

- Pavement Temperature (it will be different than air temperature)
- Parking lot area (or drive lane distance) = Length $x$ Width
- Amount of material your truck or sander delivers at each setting and speed.


## TIPS:

- De-icers melt snow and ice. They provide no traction on top of snow and ice.

Use less! About one tsp. of salt
contaminates 5 gallons of water.

- Anti-icing prevents the bond from forming between pavement and ice.
- De-icing works best if you plow before applying material.
- Pick the right material for the pavement temperatures.
- Sand only works on top of snow as traction. It provides no melting.
- Anti-icing chemicals must be applied prior to snow fall.
- NaCl (road salt) does not work on cold days, less than $15^{\circ} \mathrm{F}$.

Melt Times for Salt ( NaCl ) at Different Pavement Temperatures

| Pavement Temp. ${ }^{\circ} \mathrm{F}$ | One Pound of Salt (NaCl) melts | Melt Times |
| :---: | :---: | :---: |
| $30^{\circ}$ | 46.3 lbs of ice | 5 min. |
| $25^{\varrho}$ | 14.4 lbs of ice | 10 min. |
| $20^{\varrho}$ | 8.6 lbs of ice | 20 min. |
| $15^{\varrho}$ | 6.3 lbs of ice | 1 hour |
| $10^{\varrho}$ | 4.9 lbs of ice | Dry salt is ineffective and will blow away be- <br> fore it melts anything |

Pick your material based on lowest practical melting temperature, not eutectic temperature which is often listed on the bag.

| Melting Characteristics |  |
| :--- | :--- |
| Chemical | Lowest Practical Melting Temp. |
| $\mathrm{CaCl}_{2}$ (Calcium Chloride) | $-20^{\circ} \mathrm{F}$ |
| KAc (Potassium Acetate) | $-15^{\circ} \mathrm{F}$ |
| $\mathrm{MgCl}_{2}$ (Magnesium Chloride | $-10^{\circ} \mathrm{F}$ |
| NaCl (Sodium Chloride) | $15^{\circ} \mathrm{F}$ |
| CMA (Calcium Magnesium Acetate) | $20^{\circ} \mathrm{F}$ |
| Blends | Check with manufacturer |
| Winter Sand/Abrasives | Never melts-provides traction only |


| Increase rate: | Decrease Rate: |
| :--- | :--- |
| Compaction occurs \& cannot be removed mechanically | Light snow or light freezing rain |
| There is a lot of snow left behind | Pavement temperature is rising <br> Subsequent applications |

Minnesota Pollution Control Agency

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File available at www.pca.state.mn.us/roadsalt

## Deicing Application Rate Guidelines for Parking Lots and Sidewalks

These rates are adapted from road application guidelines (Mn Snow \& Ice Control Field Handbook, Manual 2005-1). Develop your own application rates using the guidelines as a starting point and modify them incrementally over time to fit your needs. The area should first be cleared of snow prior to applying chemical.

| Pavement Temp. ( ${ }^{\circ} \mathrm{F}$ ) and Trend $(\uparrow \downarrow)$ | Weather Condition | Maintenance Actions | Application Rate in lbs. per 1000 square foot area |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | Salt <br> Prewetted/ Pretreated With Salt Brine | Salt Prewetted/ Pretreated With Other Blends | Dry Salt | Winter Sand (abrasives) |
| $>30 \cdot \underline{-} \uparrow$ | Snow | Plow, treat intersections only | 0.75 | 0.5 | 0.75 | not recommended |
|  | Frz. Rain | Apply chemical | 1.25 | 1.0 | 1.5 | not recommended |
| $30^{\circ} \downarrow$ | Snow | Plow \& apply chemical | 1.25 | 1.0 | 1.5 | not recommended |
|  | Frz. Rain | Apply chemical | 1.5 | 1.25 | 1.75. | not recommended |
| 25-30 $\uparrow$ | Snow | Plow \& apply chemical | 1.25 | 1.0 | 1.5 | not recommended |
|  | Frz. Rain | Apply chemical | 1.5 | 1.25 | 1.75 | not recommended |
| 25-30ㄴ $\downarrow$ | Snow | Plow \& apply chemical | 1.25 | 1.0 | 1.5 | not recommended |
|  | Frz. Rain | Apply chemical | 1.75 | 1.5 | 2.25 | 3.25 |
| 20-25 $\uparrow$ | Snow or Frz. Rain | Plow \& apply chemical | 1.75 | 1.5 | 2.25 | $\begin{gathered} \hline 3.25 \text { for frz. } \\ \text { rain } \\ \hline \end{gathered}$ |
| 20-25¢ | Snow | Plow \& apply chemical | 2.0 | 2.0 | 2.75 | not recommended |
|  | Frz. Rain | Apply chemical | 2.5 | 2.0 | 3.0 | 3.25 |
| $\begin{gathered} 15^{\circ} \text { to } 200^{\circ} \\ \uparrow \uparrow \end{gathered}$ | Snow | Plow \& apply chemical | 2.0 | 2.0 | 2.75 | not recommended |
|  | Frz. Rain | Apply chemical | 2.5 | 2.0 | 3.0 | 3.2 |
| $\begin{gathered} 15^{\circ} \text { to } 20^{\circ} \\ \downarrow \end{gathered}$ | Snow or Frz. Rain | Plow \& apply chemical | 2.5 | 2.0 | 3.0 | $\begin{aligned} & 3.25 \text { for frz. } \\ & \text { rain } \end{aligned}$ |
| $\begin{gathered} 0 \text { to } 15^{\circ} \uparrow \\ \downarrow \end{gathered}$ | Snow | Plow, treat with blends, sand hazardous areas | not recommended | 3.0 | not recommended | 5.0 spot treat as needed |
| $<0^{\circ}$ | Snow | Plow, treat with blends, sand hazardous areas | not recommended | 4.5 | not recommended | 5.0 spot treat as needed |

To determine the amount of material needed, take the application rate x parking lot area / $1000 \mathrm{ft}^{2}$. Example: Given a 300,000 sq. ft. parking lot and an application rate of $1.5 \mathrm{lbs} / 1000 \mathrm{ft}^{2} \quad 1.5 \times 300,000=450,000 \quad 450,000 / 1000=450 \mathrm{lbs}$ (nine 50 lb . bags).

| Anti-Icing Guidelines |  |  |  |
| :--- | :---: | :---: | :---: |
| These are a starting point only. Adjust based on your experience. |  |  |  |
| Condition | Gallons/1000 sq. ft . |  | Other Products |
|  | $\mathbf{M g C l}_{2}$ | Salt Brine |  |
|  | $0.2-0.4$ | $0.3-0.6$ |  |
| 2. Prior to frost or black ice event | $0.2-0.4$ | $0.3-0.8$ |  |
| 3. Prior to light or moderate snow | $0.2-0.4$ | 0.3 |  |

CAUTION: Too high an application rate may result in slippery conditions or tracking.

